## I. Simon

an derive the quantity

is approximately constant and h slowly, but this may be within

FUNCTION OF  $T/\phi$ 

 $(U-U_0)/T$  0.010 0.017 0.043 0.091 0.158 0.248 0.359 0.491 0.645

ntegrating the relationship

(3)

depends only on volume. The , V, T values on the melting in figure 9, together with the is.

be extrapolated to 0°K with ationship at absolute zero can m this in turn the compressiund and is shown in table 5. res may also be obtained.

HELIUM AT 0°K

pressure	$10^5 \beta$
(atm)	(atm-1)
295	54
200	76
136	103
88	140
50	190

Thermodynamic properties and melting of solid helium

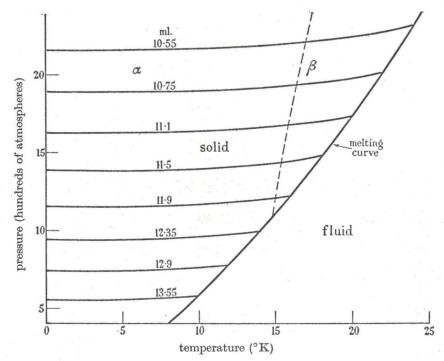


FIGURE 9. The isochores of solid helium at eight molar volumes.

----, transition line in the solid.

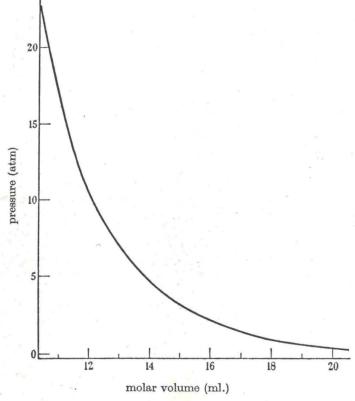


FIGURE 10. The pressure-volume relation in solid helium at 0°K.